CLAIMS

I claim:

1. A fluorophosphate glass formed from a composition comprising on a mol basis:

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a metaphosphate, Ba(PO3)2, from 10 to 60 percent; a metaphosphate, Al(PO3)3, from 10 to 60 percent; a fluoride, BaF2, from 10 to 75 percent; and a rare earth dopant.
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2. A fluorophosphates glass formed from a composition comprising on a mol basis:

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a metaphosphate, Ba(PO3)2, from 10 to 60 percent; a metaphosphate, Al(PO3)3, from 10 to 60 percent; a fluoride, RFn, from 10 to 75 percent; and a dopant.
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- 3. The glass as in claim 2 wherein the fluoride is one of the group of: BaF2, CaF2, MgF2, PbF2 and BiF3.
- 4. The glass as in claim 2 wherein the dopant is selected from the group of:

the rare earth elements neodymium (Nd), erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium (Ho), praseodymium (Pr); an oxide of manganese (Mn); and mixtures thereof.

5. The glass as in claim 4 wherein the dopant is selected from the oxides of the rare earth elements.

- 6. The glass as in claim 4 wherein the dopant on a mol percent basis is 2 to 15 percent.
- 7. The glass as in claim 4 wherein the dopant is selected from the fluorides of the rare earth elements.
- 8. A fluorophosphate glass formed from a composition comprising on a mol basis:
 - a metaphosphate, Ba(PO3)2, from 10 to 60 percent;
 - a metaphosphate, Al(PO3)3, from 10 to 60 percent;
 - a fluoride, BaF2, from 10 to 75 percent; and
 - a dopant, from 2 to 15 percent, selected from the group of:

the oxides of the rare earth elements neodymium (Nd), erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium (Ho), praseodymium (Pr);

an oxide of manganese (Mn); and mixtures thereof.

9. A fluorophosphate glass formed from a composition comprising on a mol basis:

a metaphosphate, Ba(PO3)2, from 0 to 100 percent;

a metaphosphate, Al(PO3)3, from 0 to 100 percent;

a fluoride, BaF2, from 5 TO 30 percent; and

a dopant, from 2 to 15 percent, selected from the group of:

the oxides of the rare earth elements neodymium (Nd), erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium (Ho), praseodymium (Pr);

an oxide of manganese (Mn); and mixtures thereof.

10. A fluorophosphate glass formed from a composition comprising on

a mol basis:

a metaphosphate, Ba(PO3)2, from 0 to 45 percent;
a metaphosphate, Al(PO3)3, from 5 to 30 percent;
a fluoride, BaF2, from 45 to 75 percent; and
a dopant, from 2 to 15 percent, selected from the group of:
the oxides of the rare earth elements neodymium (Nd),
erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium
(Ho), praseodymium (Pr);
an oxide of manganese (Mn); and mixtures thereof.

11. A fluorophosphate glass formed from a composition comprising on a mol basis:

a metaphosphate, Ba(PO3)2, approximately 10 percent; a metaphosphate, Al(PO3)3, approximately 18 percent; a fluoride, BaF2, approximately 70 percent; and a dopant, approximately 2 percent: of the oxide of neodymium (Nd).

12. A fluorophosphate glass formed from a composition comprising on a mol basis:

a metaphosphate, Ba(PO3)2, approximately 10 percent; a metaphosphate, Al(PO3)3, approximately 18 percent; a fluoride, BaF2, approximately 70 percent; and a dopant, approximately 2 percent: of the oxide of erbium (Er)..

13. A method for making fluorophosphates glass comprising the steps of:

batching the glass components;
melting the glass components to form a molten mixture;

cooling the molten glass mixture to a solid state; annealing the glass in the solid state;

slowly cooling the annealed glass to approximately ambient temperature;

the glass components comprised on a mol percent basis of Ba(PO3)2 from 10 to 60 percent; Al(PO3)3 from 10 to 60 percent; a fluoride selected from the group of BaF2, CaF2, MgF2, PbF2 and BiF3 from 10 to 75 percent; and a dopant from 2 to 15 percent on a mol percent basis selected from the group of Nd2O3, Er2O3, Yb2O3, Tm2O3, Tb2O3, Ho2O3, Pr2O3 and MnO and mixtures thereof.

- 14. The method as in claim 13 wherein the melting of the glass is performed in the temperature range of 1,200°C to 1,250°C in platinum crucibles in a dry argon atmosphere for from 4 to 5 hours.
- 15. The method as in claim 13 wherein the annealing of the glass is performed in the temperature range of 320°C to 340°C for from 8 to 10 hours.
- 16. A fluorophosphate glass formed from a composition comprising on a mol basis:
 - a metaphosphate, Ba(PO3)2, from 10 to 60 percent; a metaphosphate, Al(PO3)3, from 10 to 60 percent; a fluoride, BaF2 + RFx, from 10 to 90 percent; and a dopant.
- 17. The glass as in claim 16 wherein the fluoride RFx is one of the group of:
 - CaF2, MgF2, PbF2 and BiF3.
 - 18. The glass as in claim 16 wherein the dopant is selected from the

group of:

the rare earth elements neodymium (Nd), erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium (Ho), samarium (Sm), europium (Eu), praseodymium (Pr); an oxide of manganese (Mn); and mixtures thereof.

- 19. The glass as in claim 18 wherein the dopant is selected from the oxides of the rare earth elements.
- 20. The glass as in claim 18 wherein the dopant on a mol percent basis is 2 to 15 percent.
- 21. The glass as in claim 18 wherein the dopant is selected from the fluorides of the rare earth elements.
- 22. A fluorophosphate glass formed from a composition comprising on a mol basis:
 - a metaphosphate, Ba(PO3)2, from 5 to 60 percent;
 - a metaphosphate, Al(PO3)3, from 5 to 60 percent;
- a fluoride, BaF2 + RFx selected from the group of CaF2, MgF2, PbF2 and BiF3, from 10 to 90 percent; and
 - a dopant, from 2 to 20 percent, selected from the group of:
- the oxides of the rare earth elements neodymium (Nd), erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium (Ho), praseodymium (Pr), samarium (Sm), europium (Eu);
 - an oxide of manganese (Mn); and mixtures thereof.
- 23. A fluorophosphate glass formed from a composition comprising on a mol basis:
 - a metaphosphate, Ba(PO3)2, from 0 to 100 percent;
 - a metaphosphate, Al(PO3)3, from 0 to 100 percent;

a fluoride, BaF2 + RFx selected from the group of CaF2, MgF2, PbF2 and BiF3, from 5 to 90 percent; and

a dopant, from 2 to 20 percent, selected from the group of:

the oxides of the rare earth elements neodymium (Nd), erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium (Ho), praseodymium (Pr), samarium (Sm), europium (Eu);

an oxide of manganese (Mn); and mixtures thereof.

24. A fluorophosphate glass formed from a composition comprising on a mol basis:

a metaphosphate, Ba(PO3)2, from 0 to 45 percent;

a metaphosphate, Al(PO3)3, from 5 to 30 percent;

a fluoride, BaF2 + RFx selected from the group of CaF2, MgF2, PbF2 and BiF3, from 45 to 90 percent; and

a dopant, from 2 to 20 percent, selected from the group of:

the oxides of the rare earth elements neodymium (Nd), erbium (Er), ytterbium (Yb), thulium (Tm), terbium (Tb), holmium (Ho), praseodymium (Pr), samarium (Sm), europium (Eu);

an oxide of manganese (Mn); and mixtures thereof.

25. A fluorophosphate glass formed from a composition comprising on a mol basis:

a metaphosphate, Ba(PO3)2, approximately 10 percent;

a metaphosphate, Al(PO3)3, approximately 18 percent;

a fluoride, BaF2 + RFx selected from the group of CaF2, MgF2,

PbF2 and BiF3, approximately 90 percent; and

a dopant, approximately 5 percent:

of the oxide of neodymium (Nd).

26. A fluorophosphate glass formed from a composition comprising on

a mol basis:

a metaphosphate, Ba(PO3)2, approximately 10 percent;
a metaphosphate, Al(PO3)3, approximately 18 percent;
a fluoride, BaF2 + RFx selected from the group of CaF2, MgF2,
PbF2 and BiF3, approximately 90 percent; and
a dopant, approximately 10 percent:
of the oxide of erbium (Er).

27. A method for making fluorophosphates glass comprising the steps of:

batching the glass components;
melting the glass components to form a molten mixture;
cooling the molten glass mixture to a solid state;
annealing the glass in the solid state;

slowly cooling the annealed glass to approximately ambient temperature;

the glass components comprised on a mol percent basis of Ba(PO3)2 from 10 to 60 percent; Al(PO3)3 from 10 to 60 percent; a fluoride of BaF2 + RFx where RFx is selected from the group of, CaF2, MgF2, PbF2 and BiF3 from 10 to 90 percent; and a dopant from 2 to 20 percent on a mol percent basis selected from the group of Nd2O3, Er2O3, Yb2O3, Tm2O3, Tb2O3, Ho2O3, Pr2O3, Sm2O3, Eu2O3 and MnO and mixtures thereof.

- 28. The method as in claim 13 wherein the melting of the glass is performed in the temperature range of 1,200°C to 1,250°C in vitreous carbon in a dry argon atmosphere for from 4 to 5 hours.
- 29. The method as in claim 13 wherein the annealing of the glass is performed in the temperature range of 320°C to 340°C for from 8 to 10 hours.